

CONCISE SYNTHESIS OF BELAMCANDAQUINONES A AND B BY PALLADIUM (0) CATALYZED CROSS-COUPLING REACTION OF BROMOQUINONE WITH ARYLBORONIC ACIDS

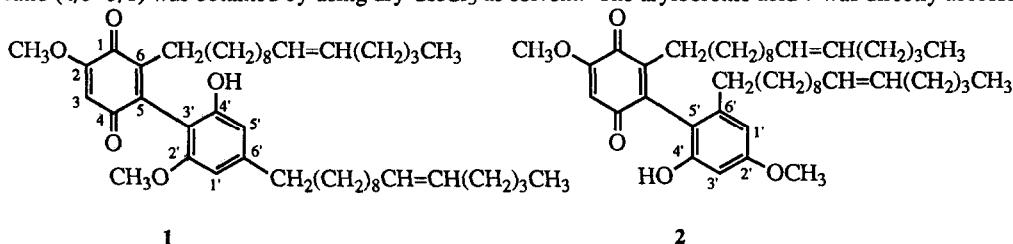
Yoshiyasu Fukuyama, * Yuuko Kiriya, and Mitsuaki Kodama

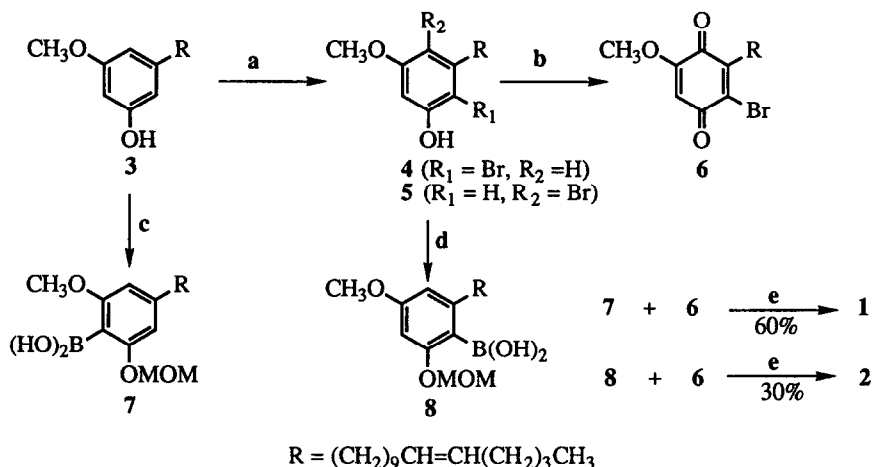
Faculty of Pharmaceutical Sciences, Tokushima Bunri University, Yamashiro-cho, Tokushima 770, Japan

Abstract: Belamcandaquinones A (**1**) and B (**2**), new dimeric 1,4-benzoquinone derivatives with cyclooxygenase inhibitory activity have been synthesized by palladium (0) catalyzed cross-coupling reaction between bromoquinone and arylboronic acids readily derived from belamcandol B (**3**), their common biogenetic precursor.

Belamcandaquinones A (**1**) and B (**2**) were isolated as specific cyclooxygenase inhibitor from a medicinal plant, *Belamcanda chinensis* and their structures elucidated on the basis of spectral data and some chemical evidence.¹ Since these compounds were supposed to be biosynthesized via an oxidative coupling of belamcandol B (**3**), a co-metabolite in this plant, the same length and location of the double bond as in **3** were allocated to the two long alkenyl side chains involved in **1** and **2**. In order to confirm the proposed structures of **1** and **2** and to obtain sufficient amount of these compounds for further biological studies, we attempted the synthesis of **1** and **2** starting from **3**, regarded as an important intermediate in their biosyntheses. Here we report facile synthesis of **1** and **2** realized by the successful use of bromobenzoquinone in the palladium catalyzed cross-coupling reaction with arylboronic acids.

In the retrosynthetic analysis of both **1** and **2** the key carbon-carbon bond between the benzoquinone nucleus and the aryl groups can be derived by a palladium-catalyzed Suzuki reaction³ from 5-bromo-ardisianone (**6**) and phenylboronic acids **7** or **8**.⁴ Three components **6**, **7**, and **8** requisite for the coupling reaction can be readily prepared from belamcandol B (**3**), which is available in a gram-scale by synthesis.⁵ The bromoquinone **6** was prepared by selective bromination with NBS followed by air oxidation. The highest ratio (4/5=5/1) was obtained by using dry CHCl₃ as solvent. The arylboronic acid **7** was directly accessed





Scheme a. NBS, CHCl_3 , rt (80%); b. O_2 , Salcomine, DMF, rt (75%); c. 1. MOMCl, *i*-Pr₂EtN, 2. *n*-BuLi, THF, -78 °C, then $\text{B}(\text{OMe})_3$, 3. H^+ (77%); d. 1. MOMCl, *i*-Pr₂EtN, 2. *n*-BuLi, THF, -78 °C, then $\text{B}(\text{OMe})_3$, 3. H^+ (63%). e. 5 mol% Pd(PPh₃)₄, Na₂CO₃, THF, reflux, 2. 48% HBr, MeOH

by the addition of trimethyl borate to a THF solution of the lithiated species formed from the MOM-protected **3** and *n*-BuLi *in situ* at -78 °C. On the other hand, **8** was derived in high yield from **5** by lithiation with *n*-BuLi followed by the addition of trimethyl borate. The cross-coupling reaction of the boronic acid **7** with the bromobenzoquinone **6** by 5 mol% pd[PhP₃]₄ in the presence of 2M Na₂CO₃ under refluxing THF was smoothly proceeded to give the coupling product in 60% yield. In addition, applying this protocol to the cross-coupling of the other boronic acid **8** with **6**, the reaction was in competition with protodeboronation of the boronic entity to yield the desired coupling product in poor yield (30%).⁶ The coupling products thus obtained were transformed into belamcandaquinones **1** and **2**,⁷ respectively, by deprotection of the MOM group with 48% HBr.

These results demonstrate a successful use of bromobenzoquinone in the palladium catalyzed cross-coupling reaction with arylboronic acids and have unambiguously established the structures of **1** and **2**.

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References and Notes

- In the preceding paper.
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- All the spectral data of the synthetic **1** and **2** were completely identical with those of natural ones.